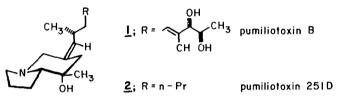
Tetrahedron Letters, Vol.26, No.35, pp 4167-4170, 1985 0040-4039/85 \$3.00 + .00 Printed in Great Britain ©1985 Pergamon Press Ltd.

THE SYNTHESIS OF INDOLIZIDINES BY INTRAMOLECULAR ENE CYCLIZATIONS. PREPARATION OF (E)-ALKYLIDENE ANALOGS OF PUMILIOTOXIN A.

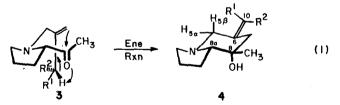
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Summary: 6-Alkylidene-8-hydroxyindolizidines 4 can be prepared by Lewis acidcatalyzed intramolecular ene cyclization of proline-derived ketones 3.

The (Z)-6-alkylideneindolizidine ring is the common structural feature of the pumiliotoxin A alkaloids,<sup>1</sup> e.g. <u>1</u> and <u>2</u>. As a result of the significant cardiotonic activity of pumiliotoxin  $B^{1,2}$  and simpler analogs,<sup>2</sup> and the scarcity of these natural products in nature,<sup>1</sup> chemical synthesis<sup>3</sup> of indolizidines

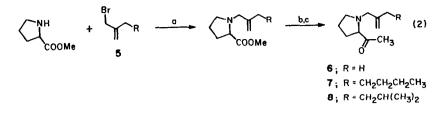


of this type is important. The 3-alkylidenecyclohexanol portion of the pumiliotoxin A alkaloids, in particular the axial nature of the C-8 hydroxyl group,<sup>1</sup> suggests that these alkaloids might be prepared by a Type II intramolecular ene reaction.<sup>4,5</sup> In this communication, we report the successful synthesis



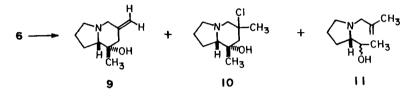
of indolizidines 4 by the intramolecular ene strategy of eq 1.

The cyclization substrates  $6-8^6$  were prepared (30-45% overall yield) from proline and the corresponding allylic bromide  $5^7$  as summarized in eq 2. Unsaturated ketone 6 was best cyclized in the presence of 2 equiv of freshly sublimed AlCl<sub>3</sub> (CH<sub>2</sub>Cl<sub>2</sub>, 25°C, 1.5 h)<sup>8</sup>. Basic workup, followed by chromatography on alumina (activity III, 9:1 hexane-ethyl acetate) gave<sup>9</sup> alkylidene indolizidine  $9^{10}$  and chloride  $10^{10}$  in yields of 27% and 39%, respectively. The stereostructure of 9, followed from the close similarity of its <sup>1</sup>H and <sup>13</sup>C NMR spectra with those of pumiliotoxin 251D (2),<sup>11</sup> see Table. On the basis of the <sup>13</sup>C NMR spectra, chloride <u>10</u> likely has the same relative stereochemistry at C-8 and C-8a, although we have no experimental evidence for the stereochemistry at

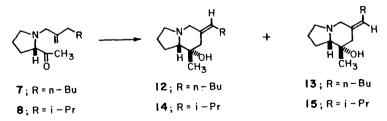


(a) K<sub>2</sub>CO<sub>3</sub>, DMF, rt; (b) LiOH (leq), THF~H<sub>2</sub>O, rt; toluene azeotrope; (c) CH<sub>3</sub>Li (leq), Et<sub>0</sub>O, 25-40°C.

C-6. A variety of other Lewis acids were ineffective (e.g.  $BF_3.OEt_2$ ,  $Me_2AlCl$ ) or afforded complex product mixtures (e.g.  $EtAlCl_2$  gave significant amounts of reduction product <u>ll</u>).



Cyclization of unsaturated ketone  $\underline{7}$  was accomplished in significantly higher yield (91% after chromatography on silica gel) under identical conditions (2 eq AlCl<sub>3</sub>, 1.5 h, 25<sup>o</sup>C) to provide a 3:1 mixture of alkylidene isomers <u>12</u> and <u>13</u>. This clean reaction demonstrates that intramolecular ene cyclization in this series is highly diastereoselective in forming the new C-8 stereocenter and that it is not essential to use alkyl aluminum halide catalysts<sup>8</sup> to obtain tertiary alcohol products<sup>8b</sup> in high yield from Lewis acid - catalyzed ene cyclizations of ketones. The alkylidene isomers could be separated on alumina (activity III, 9:1 hexane-ethyl acetate) and the minor isomer <u>13<sup>10</sup></u> was identical with a sample of pumiliotoxin "nor-ll-methyl-237A" that we had previously prepared <sup>3b</sup> by stereospecific iminium ion-vinylsilane cyclization. The major (E)-isomer <u>12<sup>10</sup></u> showed a diagnostic upfield shift<sup>12</sup> for C-7 (6.2 ppm relative to 9) in the <sup>13</sup>C NMR spectrum.



In an identical fashion,  $\underline{8}$  was cyclized to give alkylidene indolizidines  $\underline{14}$  and  $\underline{15}$  in a 6:1 ratio, together with two uncharacterized minor products.

Compound	R <sup>1</sup>	R <sup>2</sup>	<sup>13</sup> C NMR							<sup>1</sup> H NMR <sup>b</sup>				
			C-1	C-2	C-3			C-7	C-8	C-8a	H-5a	H-5 <i>B</i>	H-10	Ref.
<u>9</u>	н	H	23.3	21.3	54.5	59.3	142.1	47.5	65.8	71.3	3.45(11)	2.62(11	4.92,4.84	
251D( <u>2</u> )	(R)-1- Me- pentyl		23.3	21.1	54.6	53.2	129.8	48.9	68.3	71.8	3.78(12)	2.34(12)	5.05(9.5)	3a
nor-11- Me-237A ( <u>13</u> )	n-Bu	н	23.3	21.1	52.8	54.6	128.0	48.9	68.4	71.7	3.80(12)	2.34(12)	5.26(8)	3Ь
<u>15</u>	i-Pr	н	23.6	21.3	54.7	53.1	129.5	48.9	68.5	71.9	3.79(12)	2.32(12)	5.10(9.8)	
<u>12</u>	н	n-Bu	23.5	21.2	54.6	61.1	128.6	41.3	68.5	72.2	3.35(11)	2.62(11)	5.44(6.4)	
<u>14</u>	н	i-Pr	23.4	. 21. 2	54.5	61.0	129.4	41.4	68.4	72.1	3.30(11)	2.59(11)	5.25(9.3)	
<u>10</u>	HC1	adduct	23.1	21.2	54.3	66.3	68.5	50.9	68.5	71.6	3.24(11)			

Table: <sup>13</sup>C and <sup>1</sup>H NMR Data for Alkylideneindolizidines 4 and Chloride 10.<sup>4</sup>

<sup>a</sup>In CDC1<sub>3</sub>; ppm rel to TMS. <sup>b</sup>Coupling constants (J) in Hz are in parentheses.

Chromatography on alumina (activity IV, 20:1 hexane-ethyl acetate) provided pure samples of  $\underline{14}^{10}$  and  $\underline{15}^{10}$  in yields of 50% and 9%, respectively. Stereochemical assignments again followed unambiguously from NMR spectra (Table). Particularly diagnostic was the upfield<sup>12</sup> shift of carbon-7 in the (E)-isomer  $\underline{14}$ , and the upfield shift of carbon-5, and the corresponding downfield shift<sup>13</sup> for hydrogen-5¢, in the (Z)-isomer  $\underline{15}$ .

The stereoselectivity of the cyclization reaction appears to increase with the steric bulk of the alkylidene side chain. This result suggests that ene cyclization of the AlCl<sub>3</sub> complex of  $3^{14}$  occurs in an essentially concerted manner.<sup>15</sup> Such a process would favor loss of the diastereotopic hydrogen which would place the alkyl group in the less sterically congested R<sup>2</sup> position of <u>3</u>. We consider a stepwise process proceeding <u>via</u> a discrete C-6 cation to be much less likely, since there is no obvious reason that such a cation should undergo stereoselective deprotonation.

In conclusion, a new synthesis of indolizidines is reported that affords, with modest stereoselectivity, (E)-alkylidene analogs of the pumiliotoxin A alkaloids. Significantly, these syntheses demonstrate that Lewis-acid catalyzed intramolecular ene reactions can be successfully employed with strongly basic substrates.

<u>Acknowledgment:</u> This study was supported by PHS Grant HL-25854 and assisted by NSF Departmental instrumentation grants. L.E.O. also aknowledges support from the Camille and Henry Dreyfus and Alexander von Humboldt Foundations.

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- 5. For earlier demonstrations of this strategy for preparing axial cyclohexanols in the terpene series, see Andersen, N.H.; Uh, H.-S.; Smith, S.E.; Wutz, P.G.M. J. Chem. Soc., Chem. Commun. 1972, 956. McCurry, P.M.; Singh, R.K. <u>Tetrahedron Lett.</u> 1973, 3325. Andersen, N.H.; Ladner, D.W. Synthetic Commun. 1978, 8, 449.
- 6. The ketones prepared in this investigation were racemic. However, the preparation of ketones of this type with high enantiomeric purity should be possible, see ref. 3c.
- 7. Halides <u>5</u> (R=n-Bu, i-Bu) were prepared in 30-40% overall yield from 2,3-dibromopropene in 4 steps: (a) RMgBr,  $Et_2O$ , refl; (b) t-BuLi (2 eq), THF, -78 <sup>O</sup>C; DMF, rt; 0.1 <u>N</u> HCl; (c) NaBH<sub>4</sub>, CeCl<sub>3</sub>, MeOH-H<sub>2</sub>O, rt; (d) Ph<sub>3</sub>P, CBr<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, refl.
- (a) For a review of the Lewis acid-catalyzed ene reactions, see Snider, B. <u>Accts. Chem. Res. 1980</u>, <u>13</u>, 426. (b) For recent examples of Lewis acid-catalyzed intramolecular ene cyclizations of ketones, see Jackson, A.C.; Goldman, B.E.; Snider, B. <u>J. Org. Chem. 1984</u>, <u>49</u>, 3988.
- 9. The ratio of <u>9</u> and <u>10</u> was extremely sensitive to the reaction conditions suggesting that 10 is a secondary product formed by HCl addition to 9.
- 10. New compounds were homogeneous by TLC analysis and showed appropriate 250 MHz <sup>1</sup>H NMR, 63 MHz <sup>13</sup>C NMR, IR and mass spectra. Molecular composition of key intermediates was confirmed by high resolution MS.
- 11. Tokuyama, T.; Daly, J.W.; Highet, R.J. Tetrahedron 1984, 40, 1183.
- 12. Stothers, J.B. "Carbon-13 NMR Spectroscopy"; Academic Press: New York, 1972; pp 112-118.
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- 14. Under the conditions employed it is likely that the nitrogen and the carbonyl oxygen are coordinated to separate molecules of AlCl<sub>3</sub>.
- 15. For a recent unambiguous demonstration of concert in a quite different Lewis acid promoted ene cyclization, see Oppolzer, W.; Mirza, S. <u>Helv.</u> <u>Chem. Acta 1984</u>, <u>67</u>, 730.

(Received in USA 6 June 1985)